

## **Lead Uptake and Lead Loss in the Fresh Water Field Crab, *Barytelphusa guerini*, on Exposure to Organic and Inorganic Lead**

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Lead is a heavy metal which is widely used in paint industry, pigments, dyes, electrical components and electronics, plastic chemicals and in various other things (Hodson et al 1984). Since some of the lead salts are soluble in water, lead presents a potential threat to aquatic organisms. However enrichment of lead within aquatic organisms and its transfer through the food chain constitutes a danger to man. Many such investigations have been directed towards studying the toxic effects of lead in fishes (Holcombe et al 1976; Reichert et al 1979; Hodson 1979). Studies dealing with invertebrates include those on mortality, growth and lead uptake in *Lymnaea palustris* and bioaccumulation of heavy metals in oysters and mussels (Borgmann et al 1978; Martincic et al 1984).

Little information exists regarding the effect of lead on the fresh water crustaceans (Anderson 1978, Gilles and Pequeux 1983). Hence the present investigation has been undertaken to study the uptake and loss of lead on exposure to subtoxic levels of organic and inorganic lead.

### **MATERIALS AND METHODS**

Fresh water crabs were collected from the local paddy fields and were acclimated to the laboratory conditions. Uninjured Males of uniform size (30-40 gms), in the intermoult stages were selected for the experiment. 10 crabs were placed in each tub containing 8 litres of water. They were fed with fish meat every day and feed was withdrawn one day prior to the experiment. After two weeks of acclimation they were exposed to different concentrations of lead acetate and lead nitrate and the  $LC_{50}$  values were determined for 96 hours according to Finney (1964).  $LC_{50}$  values for 96 hours for lead acetate was found to be 20 ppm and for lead nitrate 26 ppm, (Tulasi et al 1985). Crabs were exposed to a sublethal concentration (0.5 ppm) of lead acetate and lead nitrate for a period of 30 days. Lead content was analysed according to Kendall and Scanlon (1982) on an Atomic Absorption Spectrophotometer. (Perkin Elmer Model 373). Lead content was analysed in the hepatopancreas, muscle, gill, exoskeleton (shell) and in

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haemolymph of crab on different days of exposure (1,4,15 and 30 days). After 30 days of exposure they were transferred to normal tap water and the lead content was determined for a period of 15 days.

All glassware were thoroughly washed in 20 percent nitric acid before use and rinsed in distilled water several times. Chemicals were of analytical grade and the water used was double distilled water.

## RESULTS AND DISCUSSION

Results of the present investigation are given in Figure 1. Lead was absorbed through the gills and distributed by the haemolymph to hepatopancreas, muscle and exoskeleton. Haemolymph was found to contain the highest amount of lead followed by gill, hepatopancreas, muscle and exoskeleton.

Lead bioaccumulated over the course of study shows typical differences with a high degree of organ specificity. Such a difference in the organ specific distribution has been reported for various animals including fishes and molluscs. In the case of fishes the tissue residues of lead are highest in skeleton, gill, kidney and liver with lesser amounts in muscle and other tissues (Holcombe et al 1976; Hodson et al 1978). In molluscs the highest amount is found in kidney, gill, digestive gland, foot and mantle (Pringle et al 1968; Schulz Baldes 1974). Thus a general pattern emerges in which the kidneys, gill and digestive gland shows the highest rate of uptake while gonads, mantle and muscle take up heavy metal more slowly. On exposure to lead the fresh water crab, *Barytelphusa guerinii* was shown to accumulate lead in the following order. Haemolymph, gills, hepatopancreas, muscle, exoskeleton (shell).

In the present investigation the uptake of organic lead was found to be more in the gill and haemolymph when compared to other tissues. The studies of Olson et al (1973) on the kinetics of the uptake of organic and inorganic mercury, indicated to depend on the number of sulfhydryl groups available, polarity and also on solubility of the metal. Since the kinetics of the various forms of lead are poorly understood it might be speculated that the inorganic form of lead (Lead nitrate) is poorly soluble in cell membrane with reduced motility, accumulates in lesser amounts when compared to the organic form (lead acetate). The investigations of Hodson (1979) revealed that organic lead is more toxic than inorganic lead to the fish. The experiments of Bondy (1984) also indicate that organic lead is more toxic than inorganic lead. He suggested that the most toxic forms of lead appear to combine water solubility with hydrophobic properties. This dual nature may allow maximal penetration into the tissues and also enable a most intimate contact with membranes which also have hydrophilic and hydrophobic regions. Consequently organic lead is more toxic than inorganic lead.

When the fresh water crabs were transferred to normal tap water after 30 days of exposure the lead content was found to decrease in all the tissues and in the haemolymph. (Figure 1). Experiments

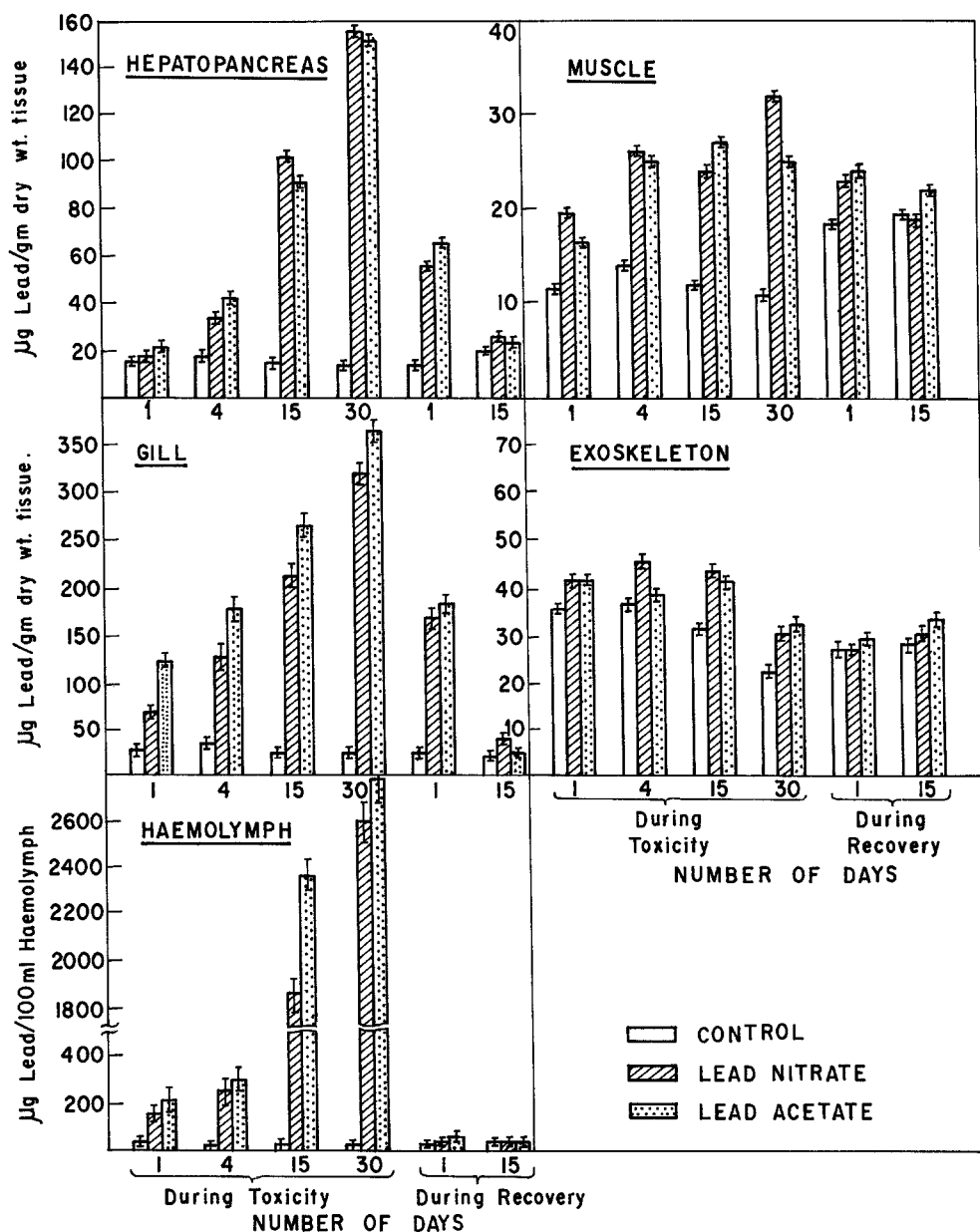


Fig. 1: Lead levels in the fresh water crab, *Barytelphusa guerini* on exposure to organic and inorganic lead. Each point is the mean of five samples.

of Holcombe et al (1976) have shown that the brook trout on transferring to normal water after 105 weeks of exposure were found to redistribute lead upto 12 weeks. Gill and kidney tissues were loosing lead while liver tissue was further accumulating lead. After 12 weeks in control water gill, kidney and liver contained significantly lower lead residues than initially. Studies of Reichert et al (1979) have shown that after 37 days of depuration lead content was found to decrease in all the tissues except kidney tissue. Therefore salmonids exhibit a strong tendency to retain lead in the kidney for at least one month after termination of the experiment. But in our studies we did not observe any redistribution of lead from one tissue to another but were loosing lead continuously on transfer to uncontaminated water.

The investigations of Cox and Anderson (1973), Tatem and Anderson (1973) have shown that the petroleum hydrocarbons which are rich in lead compounds are taken up very rapidly by crustacean tissues but retained them only for a brief period of time releasing them very rapidly reaching to undetectable levels within 1 to 7 days following exposure. On the other hand molluscs took up hydrocarbons some what more slowly but accumulated in their tissues to a considerable amount of time and released them more slowly than did crustaceans (Neff and Anderson 1974). Complete depuration by oysters may require as long as two months. Thus we have considerable evidences to show that crustaceans not only take up pollutants rapidly but also release them rapidly as is evident from the present investigation.

In the present study an attempt has been made to find out whether shell (exoskeleton) has any advantages over soft tissues as indicators for monitoring heavy metal content. But the analysis of heavy metal content in the exoskeleton of the fresh water crab did not show any significant accumulation. By comparing our findings with the data published so far on the metal content in the exoskeleton (Bryan 1964; Wright 1977) we conclude that no similarity exists in the metal content of the shell. The multiplicity and complexity of the factors which influence the metal accumulation in shells of crustaceans lead us to conclude that a great deal of information is needed before such tissues can be employed as indicators of heavy metals in the aquatic environment. On the other hand since the haemolymph lead levels directly reflect lead exposure and increases with exposure period the measurement of blood lead levels may be used as a sensitive method of determining lead toxicity in the fresh water crab.

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